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# **REARRANGED ABIETANE DITERPENOIDS FROM CLERODENDRUM MANDARINORUM**

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Five new abictane derivatives which have a commonly rearranged abietane skeleton contained a 17(15 $\rightarrow$ 16),18(4 $\rightarrow$ 3)-diabeo-abietane framework, mandarones D -H, were isolated from the stem of *Clerodendrum mandarinorum* Diels (Verbenaceae). The structures were characterized as (16S)-12,16-epoxy-11-hydroxy-17(15 $\rightarrow$ 16), 18(4 $\rightarrow$ 3)-diabeo-abieta-3,5,8,11,13-pentaene-7-one (mandarone D, 1), 12,16-epoxy-11,14-dihydroxy-17(15 $\rightarrow$ 16),18(4 $\rightarrow$ 3)-diabeo-abieta-3,5,8,11,13,15-hexaene-7-one (mandarone E, 2), 12,16-epoxy-6,11,14-trihydroxy-17(15 $\rightarrow$ 16),18(4 $\rightarrow$ 3)-diabeo-abieta-3,5,8,11,13,15-hexaene-7-one (mandarone F, 3), 12,16-epoxy-11,14-dihydroxy-6-methoxy-17(15 $\rightarrow$ 16),18(4 $\rightarrow$ 3)-diabeo-abieta-3,5,8,11,13,15-hexaene-7,one (mandarone G, 4) and 12,16-epoxy-11,14-dihydroxy-17(15 $\rightarrow$ 16),18(4 $\rightarrow$ 3)-diabeo-abieta-3,5,8,11,13,15-hexaene-1,7-dione (mandarone H, 5) respectively, mainly based on the spectral analysis and by comparison with those of closely related compounds.

Keywords: Clerodendrum mandarinorum; Verbenaceae; Rearranged abietane; Mandarone D; Mandarone E; Mandarone F; Mandarone G; Mandarone H

## INTRODUCTION

*Clerodendrum mandarinorum* Diels is an original plant of one of the traditional Chinese medicines, "Hai Tang". The crude drug has a good reputation in the treatment of infantile paralysis and apoplexy. In a previous

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paper we have reported the isolation and characterization of rearranged abietane derivatives, mandarones A-C, from C. mandarinorum Diels (Verbenaceae) [1]. Further study of the same plant material allowed the isolation of five new rearranged abietane derivatives (1-5). In this paper the structural elucidation of 1-5 is described.



#### **RESULTS AND DISCUSSION**

Repeated column chromatography of a chloroform extract of the stem of C. mandarinorum on silica gel resulted in the isolation of compounds 1-5.

Compound 1, mandarone D, obtained as orange needles, showed a molecular ion [M<sup>+</sup>] at m/z 314.1884 in the high-resolution mass spectrometry corresponding to the molecular fromula  $C_{20}H_{26}O_3$ . The NMR spectral data (Tables I and II) were almost identical with those of uncinatone [2], and the presence of a partial structure of 12,16-epoxy-17(15  $\rightarrow$  16),18(4  $\rightarrow$  3)-diabeoabieta-7-one was supported. The difference was that uncinatone has two hydroxyl groups at C-11 and C-14, but mandarone D had a hydroxyl group at C-11, because the highly deshielded resonance due to the phenoxyl proton at  $\delta$  13.73 in uncinatone disappeared and the phenoxyl proton only was observed at  $\delta$  4.82 as well as the HMBC spectrum of mandarone D showed correlations between C-7 ( $\delta$  190.4) with H-14 ( $\delta$  7.84) and H-6 ( $\delta$  6.20). The singlet signal at  $\delta$  7.84 assigned to H-14 also suggested that no

Н	1	2	3	4	5
1α	1.57 (m)	1.61 (m)	1.58 (m)	2.50 (d)	
$1\beta$	3.22 (dd)	3.30 (dd)	3.28 (dd)	4.25 (d)	
$2\alpha$	2.20 (dd)	2.25 (dd)	2.21 (dd)		2.71 (d)
$2\beta$	2.50 (m)	2.54 (m)	2.51 (m)		2.76 (d)
6	6.20 (s)	6.28 (s)			6.27 (s)
14	7.84 (s)				
$15\alpha$	3.40 (dd)				
15β	2.87 (dd)	6.51 (q)	6.55 (s)	6.58 (s)	6.39 (s)
$16\alpha$	5.10 (m)				
Me-17	1.51 (d)	2.24 (d)	2.46 (br s)	2.47 (s)	2.43 (s)
Me-18	1.91 (s) <sup>a</sup>	$1.92(s)^{b}$	$1.91(s)^{c}$	2.03 (s)	1.98 (s)
Me-19	$1.87 (s)^{a}$	$1.88 (s)^{b}$	$1.89(s)^{c}$	2.25 (s)	2.14 (s)
Me-20	1.49 (s)	1.53 (s)	1.53 (s)	1.64 (s)	1.66 (s)
OH-6	• •		$13.79 (br s)^{d}$		
OH-11	$4.82(s)^{d}$	$6.60(s)^{d}$	$4.74(s)^{d}$	5.51 (s)	$13.72 (s)^{d}$
OH-14		$13.89 (br s)^{d}$	$13.79(s)^{d}$	$113.74 (s)^{d}$	$10.11 (s)^{d}$
OMe-6				3.98 (s)	
J (Hz)					
$1\alpha, 1\beta$	13.8	13.2	13.3	13.7	
$2\alpha, 2\beta$	14.3	14.6	13.9		14.2
$15\alpha, 15\beta$	15.6				
$15\alpha, 16\alpha$	9.0				
$15\beta,16\alpha$	7.6				
15,17	1.1	1.1			
$16\alpha, 17$	7.0				

TABLE I  $\ ^{1}\text{H}\,\text{NMR}$  spectral data of compounds 1–5 (400 MHz, CDCl<sub>3</sub>, TMS as internal standard)

 $^{a,b,c}$ Assignment may be reversed.  $^{d}$ Exchangeable with  $D_2O$ .

<b>2</b> 30.2	3	4	5
30.2	20.1		
	30.1	45.4	194.0
30.6	30.8	196.5	34.8
141.2	141.0	132.4	141.5
125.6	125.5	146.3	117.2
165.9	166.0	162.1	167.2
118.8	154.2	123.2	111.9
186.8	190.5	190.2	191.6
107.2	107.4	108.6	109.6
129.5	129.8	135.6	125.6
40.1	39.8	42.4	58.4
142.1	143.8	148.9	141.8
151.7	151.6	150.5	150.1
130.2	131.4	117.3	131.2
154.8	155.1	155.5	154.9
101.8	111.6	100.9	118.8
148.6	148.5	149.8	148.7
22.9	22.7	20.2	22.7
15.6	15.2	15.7	15.1
14.4	14.2	13.7	14.7
20.7	20.8	20.7	20.8
	$\begin{array}{c} 30.2\\ 30.6\\ 141.2\\ 125.6\\ 165.9\\ 118.8\\ 186.8\\ 107.2\\ 129.5\\ 40.1\\ 142.1\\ 151.7\\ 130.2\\ 154.8\\ 101.8\\ 148.6\\ 22.9\\ 15.6\\ 14.4\\ 20.7\\ \end{array}$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$

TABLE II  $^{-13}$ C NMR spectral data of compounds 1–5 (100 MHz CDCl<sub>3</sub>, TMS as internal standard)

hydroxyl was substituted at C-14. Therefore the hydroxyl group should be substituted at C-11. The CD spectrum [254 nm (+CE), 312 nm (-CE), 335 nm (+CE)] showed that mandarone D possessed a normal abietane S-configuration at C-10 such as uncinatone [2]. Comparison of the  $\delta_{\rm H}$  of furan ring of compound I with those of teuvincenone A [3] were easily assigned (Table I) to H $\alpha$ -15, H $\beta$ -15 and H-16 indicated that the signals at  $\delta$  3.40 (J = 15.6, 9.0 Hz),  $\delta$  2.87 (J = 15.6, 7.6 Hz) and  $\delta$  5.10 (J = 9.0, 7.6 Hz). The stereochemistry of a  $\beta$ -methyl furan part was inferred by the literature [3] and biogenesis that may be rational. The structure of mandarone D was then characterized as (16S)-12,16-epoxy-11-hydroxy-17(15  $\rightarrow$  16),18(4  $\rightarrow$  3)diabeo-abieta-3,5,8,11,13-pentaene-7-one.

Compound 2, mandarone E, obtained also as orange needles, showed  $[M^+]$  at m/z 324.1360 in HREIMS corresponding to the molecular formula  $C_{20}H_{20}O_4$ . Its UV and NMR spectral data (Tables I and II) were almost in agreement with those of 1. One difference was consistent with the presence of a 2-methyl-benzo [b] furan moiety in mandarone E [ $\delta$  2.24 (3H, d,  $J_{17,15} = 1.1$  Hz, Me-17), 6.51 (1H, q,  $J_{15,17} = 1.1$  Hz, H-15);  $\delta_{\rm C}$  101.8 (C-15), 148.6 (C-16) f and 22.9 (C-17)] instead of an  $\alpha$ -methyldihydrofuran which was condensed with an aromatic ring in 1 [ $\delta$  1.51 (3H, d,  $J_{17,16\alpha} = 7.0$  Hz, Me-17), 5.10 (1H, m,  $J_{16\alpha,15\alpha} = 9.0$  Hz,  $J_{16\alpha,15\beta} = 7.6$  Hz, H-16), 3.40 (1H, dd,  $J_{15\alpha,15\beta} = 15.6 \text{ Hz}$ , H-15 $\alpha$ ), 2.87(1H, dd,  $J_{15\alpha,15\beta} = 15.6 \text{ Hz}$ ,  $J_{16\alpha,15\beta} =$ 9.0 Hz, H-15 $\beta$ )]. These NMR spectral data indicated that mandarone E must be a 15,16-dehydro derivative of 1. Another difference from 1 was the presence of a hydroxyl at C-14, which was supported by the appearance of the hydroxyl signal at  $\delta$  13.89 and disappearance of aromatic proton signal at  $\delta$  7.84 in 1. The CD spectrum [251 nm (+CE), 302 (-CE), 334 (+CE)] showed that 2 also possessed a normal abietane S-configuration at C-10. Thus the structure of mandarone E was determined to be 12,16epoxy-11, 14-dihydroxy-17(15  $\rightarrow$  16), 18(4  $\rightarrow$  3)-diabeo-abieta-3, 5, 8, 11, 13, 15hexaene-7-one.

Compound 3, mandarone F, obtained as orange rectangles, showed  $[M^+]$ at m/z 340.1312 in HREIMS corresponding to the molecular formula  $C_{20}H_{20}O_5$ . The <sup>1</sup>HNMR spectrum was almost the same as those of 2. Mandarone F had another hydroxyl group at C-6 compared with 2, which was supported by appearance of the hydroxyl signal at  $\delta$  7.12 and disappearance of an olefinic proton at  $\delta$  6.28 assigned to H-6 in 2. The CD spectrum [252 (+CE), 288 (-CE), 321 (+CE)] showed that 3 also possessed a normal abietane S-configuration at C-10. Therefore the structure of mandarone F was characterized as 12,16-epoxy-6,11,14-trihydroxy-17 (15  $\rightarrow$  16),18(4  $\rightarrow$  3)-diabeo-abieta-3,5,8,11,13,15-hexaene-7-one.

Compound 4, mandarone G, obtained as brown column crystals, showed  $[M^+]$  at m/z 368.1258 in HREIMS corresponding to the molecular formula  $C_{21}H_{20}O_6$ . The <sup>1</sup>H NMR spectrum revealed the existence of a partial structure of 12,16-epoxy-11,14-dihydroxy-17(15 $\rightarrow$ 16),18(4 $\rightarrow$ 3)-diabeo-abieta-8,11,13-trienc-7-one and was fundamentally the same as those of 3. However, there were still some differences from 3. Mandarone G possessed two ketone groups ( $\delta_{\rm C}$  196.5, 190.2) and one methylene group [ $\delta$  2.50 (1H, d, J = 13.7 Hz) and 4.25 (1H, d, J = 13.7 Hz)] in the structure. One of the ketones could be substituted as C-1 or C-2. Owing to the strong down-field shift of the methylene group, the proton signal at  $\delta$  4.25 must be due to the methylene group substituted at C-1 and the HMBC spectrum showed that the C-5 ( $\delta$  162.1) had correlation with H-1 $\alpha$  ( $\delta$  2.50, 1H, d) and H-1 $\beta$ ( $\delta$  4.25, 1H, d). Therefore another ketone group must be at C-2. The partial structure of 12,16-epoxy-11,14-dihydroxy-17(15  $\rightarrow$  16),18(4  $\rightarrow$  3)diabeo-abieta-8,11,13-triene-2.7-dione thus deduced is corresponding to that of teuvincenone F [4], and the NMR spectral data of mandarone G was very similar to those of teuvincenone F. The <sup>1</sup>HNMR spectrum showed the presence of a methoxyl group at  $\delta$  3.98. Considering the absence of an olefinic proton and the presence of two hydroxyl groups at C-11 and C-14, the methoxyl group was rationally substituted at C-6. The absolute configuration was ascertained by CD spectral data in comparison with those of (16R)-plectrionon A [5.6]. The Cotton effect at 257 (--CE), 282 (+CE) and 345 nm (+CE) showed that mandarone G possessed the same S-configuration at C-10 as (16R)-plectrionon A. Therefore the structure of mandarone G was concluded to be 12,16-epoxy-11,14-dihydroxy-6-methoxy- $17(15 \rightarrow 16), 18(4 \rightarrow 3)$ -diabeo-abieta-3,5,8,11,13,15-hexaene-2,7-dione.

Compound 5, mandarone H, obtained as yellowish needles, showed  $[M^+]$  at m/z 338.1152 in HREIMS corresponding to the molecular formula  $C_{20}H_{18}O_5$ . Mandarone H had two ketone groups in the structure like 4, but the position was different from that of 4. In the <sup>1</sup>H NMR spectrum, the methylene proton signals were observed at  $\delta$  2.71 (1H, d, J = 14.2 Hz) and 2.76 (1H, d, J = 14.2 Hz) in mandarone H, however, the corresponding signals were at  $\delta$  2.50 and 4.25 in 4. The HMBC spectrum showed that one of ketones ( $\delta_C$  194.0) had long-range correlations with a methyl  $\delta$  1.66 (3H, s, Me-20), a methylene proton  $\delta$  2.71 (H-2 $\alpha$ ) and 2.76 (H-2 $\beta$ ). The methine proton ( $\delta$  6.27) had long-range correlations with another ketone ( $\delta$  191.6, C-7) and C-5 ( $\delta$  167.2), the remaining methine proton  $\delta$  6.39 (1H, s, H-15) had correlations with C-13 ( $\delta$  131.2) and C-16 ( $\delta$  148.7). Further long-range coupling between a methyl group  $\delta$  2.43 (3H, s, Me-17) and C-16 ( $\delta$  148.7) was also observed. These results supported the ketone groups to be located

at C-1 and C-7. The stereochemistry at C-10 was determined by CD spectrum. The negative Cotton effect at 295 and 306 nm showed that the methyl group at C-10 was  $\beta$ -oriented and that the ketone groups were substituted at C-1 and C-7 of an abietane diterpene [7]. Thus the structrue of mandarone H was drawn as 12,16-epoxy-11,14-dihydroxy-17(15  $\rightarrow$  16),18(4  $\rightarrow$  3)-diabeo-abieta-3,5,8,11,13,15-hexaene-1,7-dione.

## **EXPERIMENTAL SECTION**

### **General Experimental Procedures**

Melting points were determined on a  $X_4$  apparatus and are uncorrected. UV spectra were taken on Shimadzu UV-265 spectrometer. IR spectra were taken on Impact-410 spectrometer. <sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded on Bruder AM-400 Spectrometer. Mass spectra were measured on Finnigan MAT SSQ 710 spectrometer.

#### **Plant Material**

Stems of *C. mandarinorum* Diels cultivated in Lushan Plant Garden of Jiangxi, China were used in the present study.

## **Extraction and Isolation**

The naturally dried and pulverized stems of *C. mandarinorum* (5 kg) were extracted with hot EtOH three times. The solvent was removed in vacuum to yield crude extract (180 g). A certain volume of  $H_2O$  was added to the crude extract and the mixture was agitated thoroughly to form a suspension, which was extracted with CHCl<sub>3</sub>. After evaporation, the CHCl<sub>3</sub> extract (80 g) was subjected to column chromatography on silica gel eluting with CHCl<sub>3</sub> to give 1 (16 mg), and eluting with an EtOAc-petroleum ether gradient to give 2 (10 mg), 3 (13 mg), 4 (11 mg) and 5 (8 mg).

Compound 1 [mandarone D, (16S)-12,16-epoxy-11-hydroxy-17(15  $\rightarrow$  16), 18(4  $\rightarrow$  3)-diabeo-abieta-3,5,8,11,13-pentaene-7-one]. m.p. 216-217°C (CHCl<sub>3</sub>), yellowish needles. [ $\alpha$ ]<sub>D</sub> +51.2 (CHCl<sub>3</sub>, c, 0.0158), CD: 227(0), 254(+4.74), 289(0), 312(-1.82), 328(0), 335(+3.39) nm, IR  $\nu$  <sup>KBr</sup><sub>max</sub>: 3320(OH), 1640, 1620, 1600, 1580, 1480, 1360, 1060 cm<sup>-1</sup>, UV  $\lambda$ <sup>MeOH</sup><sub>max</sub>: 258, 314, 373 nm. <sup>1</sup>H NMR and <sup>13</sup>C NMR spectral data are listed in Tables I and II. HREIMS *m/z*: 314.1884 for C<sub>20</sub>H<sub>26</sub>O<sub>3</sub> (calcd. 314.1883).

Compound **2** [mandarone E, 12,16-epoxy-11,14-dihydroxy-17(15 $\rightarrow$ 6), 18(4 $\rightarrow$ 3)-diabeo-abieta-3,5,8,11,13,15-hexaene-7-one]. m.p. 223–224°C (CHCl<sub>3</sub>), orange needles, [ $\alpha$ ]<sub>D</sub> +75.5 (CHCl<sub>3</sub>, c, 0.0171), CD: 213 (-6.76), 228(0), 251(+4.75), 268(0), 274(-0.61), 280(0), 302(-2.32), 312(0), 334 (+3.22) nm, IR  $\nu_{\text{max}}^{\text{KBr}}$ : 3400(OH), 1600 (ketone), 1610, 1575, 1480, 1320, 1010 cm<sup>-1</sup>, UV  $\lambda_{\text{max}}^{\text{MeOH}}$ : 258, 314, 373 nm. <sup>1</sup>H and <sup>13</sup>CNMR spectral data are shown in Tables I and II. HREIMS *m*/*z*: 324.1360 for C<sub>20</sub>H<sub>20</sub>O<sub>4</sub> (calcd. 324.1362).

Compound **3** [mandarone F, 12,16-epoxy-6,11,14-trihydroxy-17(15  $\rightarrow$  16),18(4  $\rightarrow$  3)-diabeo-abieta-3,5,8,11,13,15-hexaene-7-one]. m.p. 221–222°C (CHCl<sub>3</sub>), orange rectangles [ $\alpha$ ]<sub>D</sub> +83.2 (CHCl<sub>3</sub>, c, 0.0183), CD: 211(-4.41), 231(0), 252(+4.09), 272(0), 288(-2.10), 301(0), 321(+4.48) nm, IR  $\nu_{\text{max}}^{\text{KBr}}$ : 3200(OH), 1640 (ketone), 1600, 1580, 1470, 1340, 1000 cm<sup>-1</sup>. <sup>1</sup>H and <sup>13</sup>C NMR spectral data are listed in Tables I and II, respectively. HREIMS *m*/*z*: 340.1312 for C<sub>20</sub>H<sub>20</sub>O<sub>5</sub> (calcd. 340.1311).

Compound 4 [mandarone G, 12,16-epoxy-11,14-dihydroxy-6-methoxy-17(15  $\rightarrow$  16),18(4  $\rightarrow$  3)-diabeo-abieta-3,5,8,11,13,15-hexaene-2,7-dione]. m.p. 331°C (CHCl<sub>3</sub>), brown column crystals, [ $\alpha$ ]<sub>D</sub> +45.1 (CHCl<sub>3</sub>, c, 0.0171), CD: 210(+0.49), 228(+7.68), 242(0), 257(-5.74), 275(0), 282(+1.81), 307(+0.81), 345(+3.36) nm, IR  $\nu_{max}^{\text{KBr}}$ : 3400 (OH), 1660 (ketone), 1640, 1620, 1600, 1570, 1460, 1290, 1030 cm<sup>-1</sup>. <sup>1</sup>H and <sup>13</sup>C NMR spectral data are listed in Tables I and II, respectively. HREIMS *m/z*: 368.1258 for C<sub>21</sub>H<sub>20</sub>O<sub>6</sub> (calcd. 368.1260).

Compound 5 [mandarone H, 12,16-epoxy-11,14-dihydroxy-6-methoxy-17(15  $\rightarrow$  16),18(4  $\rightarrow$  3)-diabeo-abieta-3,5,8,11,13,15-hexaene-1,7-dione]. m.p. 218-220°C (CHCl<sub>3</sub>), yellowish needles. CD: 272(0), 295(-12.3), 306 (-8.24), 331(0), 348(+4.17) nm, IR  $\lambda_{max}^{KBr}$ : 3300 (OH), 1640 (ketone), 1610, 1575,1480, 1320, 1010 cm<sup>-1</sup>, UV  $\lambda_{max}^{MeOH}$ : 237, 282 nm. <sup>1</sup>H and <sup>13</sup>C NMR spectral data are listed in Tables I and II, respectively. HREIMS *m/z*: 338.1152 for C<sub>20</sub>H<sub>18</sub>O<sub>5</sub> (calcd. 338.1154).

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